

## <u>Across</u>

**3.** figuring out how old an object is

**4.** the amount of time it takes for a certain atom to decay to half the amount it started as

**5.** a process in which an atom will go through to make its nucleus stable

**8.** the sum of the protons and neutrons in an atom's nucleus

**9.** a group of atoms that all have the same number of protons

**10.** a small particle that is inside of an atom, it has no charge

**11.** a visual example of a system that shows its known properties

12. the center of an atom that holds the protons and neutrons13. a small particle that orbits the nucleus of an atom, it has a negative charge

14. the number of protons in an atom

## <u>Down</u>

 an organized chart that shows every element and their mass and atomic number
an atom with an unstable nucleus

**6.** atoms with the same number of protons and a different number of neutrons

7. a small particle that is inside of an atom, it has a positive charge