Name: $\qquad$ Date: $\qquad$ Period: $\qquad$

## Mole Crossword

## Across

5. molar mass of salt
6. molar mass of CO2
7. molar mass of K2CO3
8. The number of representative
particles contained in one mole of a
substance; equal to
$6.02 \times 10^{\wedge} 23$
particles.
9. A term used to
refer to the mass of a
mole of any substance
10. Convert 3.55
moles NaCl to
molecules.
11. Convert 3.00
moles As2S3 to grams
Down
12. Convert 8.00 x
$10^{\wedge} 20$ molecules of H2 to moles.
13. How many atoms are in 6.2 moles of aluminum?
14. molar mass of KCl
15. What is the mass
of 9.30 moles of SiH 4
16. Convert 5.3 x

1025 molecules of
CO2 to moles.
11. The amount of a
substance that
contains $6.02 \times 10^{\wedge} 23$
representative
particles of that
substance
13. molar mass of


HCl
14. How many grams
of tin are found in
3.50 moles of tin

