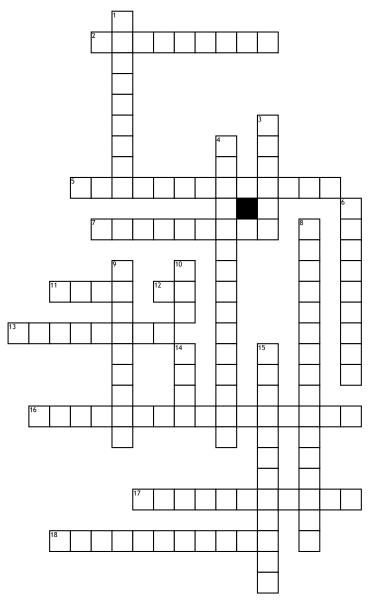
Name:	Date:
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Mole Day Crossword



Across

- 2. Clump of atoms together
- **5.** The type of math used to find the measurement of moles
- **7.** Mass of one mole in a substance
- 11. Amount of substance
- **12.** Find the molar mass of 2NH3
- **13.** The last substances, on the right of the equation

- **16.** a reactant that determines the amount of product that can be produced
- **17.** the actual amount of the substance used in percent yield
- **18.** Volume occupied by 1 mole of a gas at STP

Down

- 1. the ratio of the moles in multiple substances
- **3.** The unit of volume used in stoichiometry

- **4.** Number used to calculate molecules
- 6. This is Avogadro's number
- **8.** potential amount of substance in percent yield
- **9.** the first substances, on the left of the equation
- **10.** Find the molar mass of 2Fe2O3
- **14.** Unit of measurement that used 760 as the standard
- **15.** Ratio of actual yield to theoretical yield