

Name: \_\_\_\_\_ Date: \_\_\_\_\_

# Naming Acids and Bases

- |  |                         |
|--|-------------------------|
| 1. HCl   | A. hydrophosphoric acid |
| 2. HNO <sub>3</sub>                              | B. nitrous acid         |
| 3. NaOH  | C. chloric acid         |
| 4. H <sub>2</sub> SO <sub>3</sub>                | D. chlorous acid        |
| 5. HC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> | E. sulfuric acid        |
| 6. H <sub>2</sub> CO <sub>3</sub>                | F. sodium hydroxide     |
| 7. HBr   | G. nitric acid          |
| 8. HClO <sub>2</sub>                             | H. carbonic acid        |
| 9. H <sub>3</sub> PO <sub>4</sub>                | I. acetic acid          |
| 10. HClO <sub>3</sub>                            | J. iodic acid           |
| 11. HNO <sub>2</sub>                             | K. potassium hydroxide  |
| 12. H <sub>2</sub> SO <sub>4</sub>               | L. hydrobromic acid     |
| 13. H <sub>3</sub> PO <sub>3</sub>               | M. hydrosulfuric acid   |
| 14. H <sub>2</sub> S                             | N. hydrochloric acid    |
| 15. HIO <sub>3</sub>                             | O. perchloric acid      |
| 16. HClO <sub>4</sub>                            | P. sulfurous acid       |
| 17. Ca(OH) <sub>2</sub>                          | Q. phosphoric acid      |
| 18. H <sub>2</sub> CrO <sub>4</sub>              | R. chromic acid         |
| 19. H <sub>3</sub> P                             | S. phosphorus acid      |
| 20. KOH  | T. calcium hydroxide    |