

Name: _____

Naming Covalent Compounds

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| 1. Disilicon hexabromide | A. OF ₂ |
| 2. Phosphorous trihydride | B. N ₂ O ₅ |
| 3. Dihydrogen monoxide | C. Cl ₂ O ₇ |
| 4. Pentaphosphorous decoxide | D. PH ₃ |
| 5. Dichlorine Heptaoxide | E. CBr ₄ |
| 6. Sulfur dichloride | F. Si ₂ Br ₆ |
| 7. Carbon Dioxide | G. SF ₆ |
| 8. Dinitrogen Pentoxide | H. S ₂ CL ₃ |
| 9. Selenium Hexobromide | I. IF ₇ |
| 10. Carbon tetrabromide | J. BN |
| 11. Boron Mononitride | K. SeBr ₆ |
| 12. Oxygen Difluoride | L. CO ₂ |
| 13. Sulfur Hexafluoride | M. P ₄ O ₁₀ |
| 14. Iodine Heptaflouride | N. SCl ₂ |
| 15. Disulfur Trichloride | O. H ₂ O |