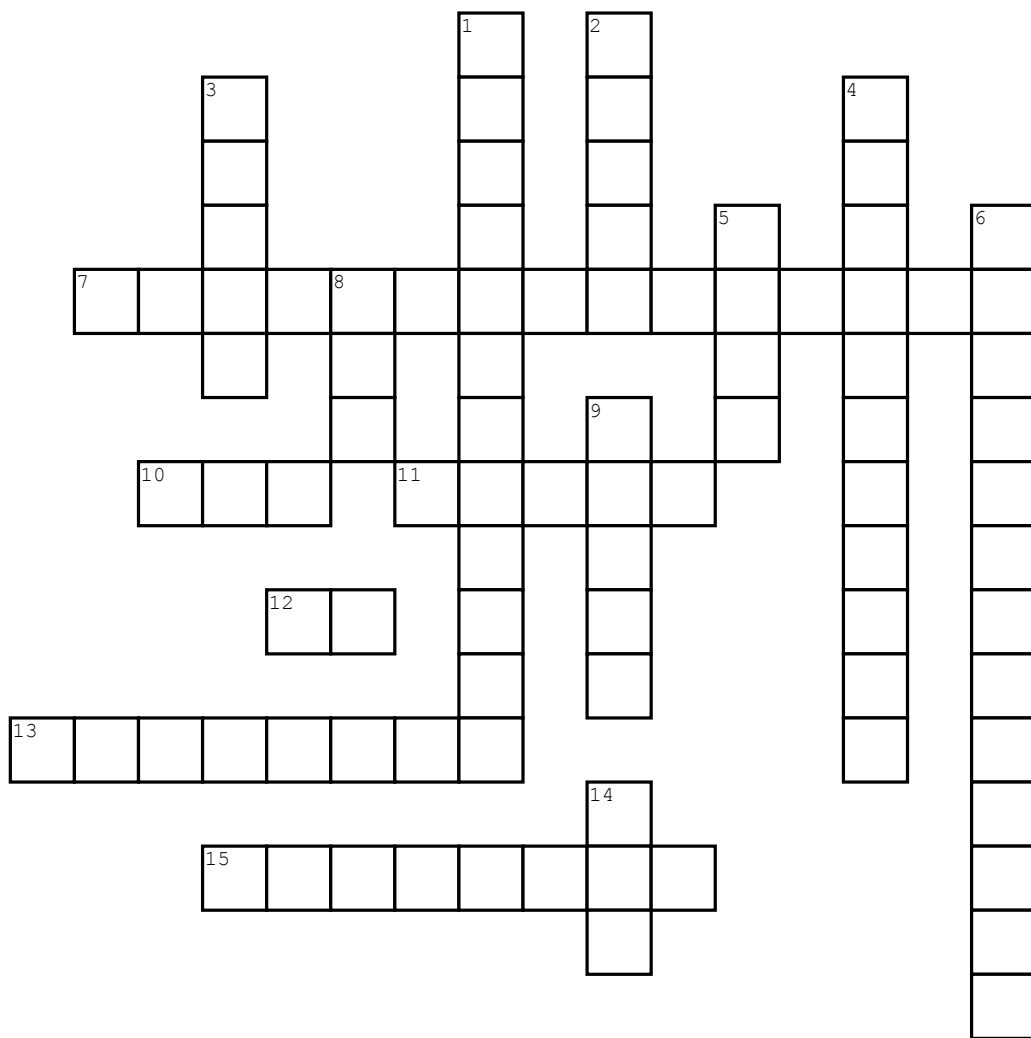


Name: _____ Date: _____ Period: _____

Nomenclature Review



Across

7. What is the name of CaF_2
 10. *extra credit* what's YOUR class number
 11. if it contains a non-metal and a metal what compound is it?
 12. What is the formula for potassium fluoride?
 13. What is made up of atoms?

15. what compound has two non-metals

Down

1. What is the correct name for the ionic compound, Cr_2O_3 ?
 2. Which element is always written first in an ionic compound the metal or the non-metal element?
 3. What is the formula for the ionic compound lead(II) chloride?

4. What is the name of Na_2O

5. What are atoms with a charge?
 6. What is the name for the ionic compound Hg_2I_2 ?
 8. what do binary compounds have to end in?
 9. What is the formula of titanium(III) bromide?
 14. What is the formula for compound manganese(II) sulfide?

Word Bank

chromic oxide	KF	PbCl_2	covalent
221	mercury iodide	Molecule	metal
TiBr_3	ied	MnS	Ions
sodium oxide	ionic	calcium fluoride	