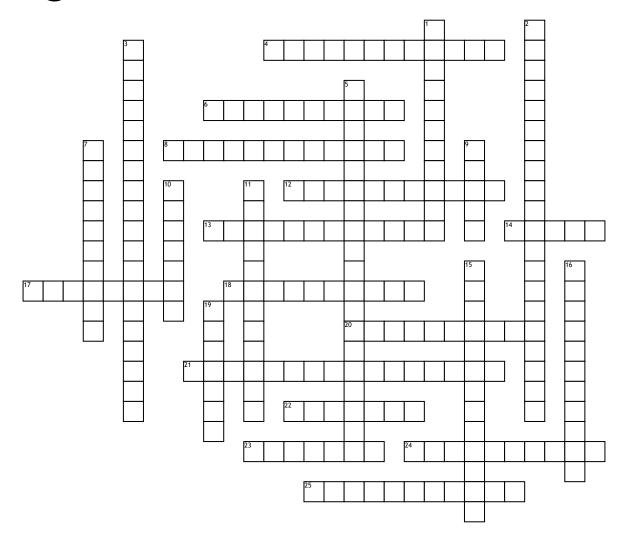
## Organic Reaction Mechanisms



## **Across**

- **4.** In dehydrohalogenation reactions, the preferred product is the alkene that has the greater number of alkyl groups attached to the double bonded carbon atoms.
- **6.** Name the first step in a radical reaction
- **8.** An atom or group atoms which breaks away from the rest of the molecule
- **12.** A molecule that contains at least one unpaired electron
- 13. Replacement of one group by another
- 14. Sn1 is given by a .....
- **17.** A class of chemical reactions in which an atom or group of atoms is added to a molecule
- **18.** A reaction that proceeds at a rate that depends on one reactant concentration

- **20.** Reactions transfer energy to the surroundings and the temperature of the surroundings increase.
- **21.** When the reaction chooses one product to produce exclusively
- **22.** A molecule that contains at least one unpaired electron
- 23. A term used to describe alkenes
- **24.** Name a factor influencing direction of substitution reaction
- **25.** The rule which states that the more substituted a carbonation is, the more stable the molecule is

## Down

- **1.** The addition of a proton or hydrogen cation
- **2.** Reaction involving a change in only one molecule of ionic structure.
- **3.** The slowest step in a mechanism is known as...

- **5.** The chemical combination of two molecular entities in a reaction that can be considered either reversible or irreversible
- **7.** The radical formation step
- 9. Sn2 is given by a ..... R group
- **10.** A substance in which other materials dissolve to form a material
- 11. Replacement of one group by another
- **15.** The removal of a proton in a acid-base reaction
- **16.** Reactions take in energy and the temperature of the surroundings decreases
- **19.** A mixture that has equal amount of enantiomers