

Name: _____

Date: _____

Periodic Table

G O K N E I B P G I H O A S B 1 D D O C T M S M
 Y U U K S G M V A E 1 V S T 3 3 S S E E P P S U
 S L F T P U K N L 2 M F O D O L Y S P F M S A I
 U 2 S U I E V 1 L R K A S 2 N M I N H K K O M S
 A 1 O C B C P I I F 3 H N G B U I H G O I D C E
 L R L G A O U V U 1 P S R O M E O C O P U I I N
 G A N 2 3 T A E M V G O L I S A R P N C M U M G
 C V M U I D N I C R U K P A S M P Y M U B M O A
 F F K N B D A E E P G T L T L F U V L O M T T M
 C Y 1 1 E V R A 1 I P R N 1 3 G K I K L C B A C
 H G S H E G C A D C R E H K F 2 C G R N I E E T
 P F U 2 T T N 3 M A M F F T K P B R E A 2 U F R
 A L 1 U I K 2 C D E G F H U V F O O U R B Y M 1
 U G R V L K S I L H Y 3 K V V Y R U 3 O L A K S
 E L I U N S U E M T S M D A I D O P K C A V D E
 M T T 3 B M S U A 3 P U O R G T N 2 B F O K B I
 Y 3 G K O I I S T R O N T I U M S A 3 L 1 1 1 T
 3 B 2 V T L D E H N S I A L U M I N I U M M N R
 A V N K L U N I K O T P L F R A N C I U M I K E
 1 D U A D K G T U 2 2 2 M U I S S A T O P Y 3 P
 K U H G E O G D 1 M P S L A T E M I L A K L A O
 U T I K S N O R T C E L E E C N E L A V 1 I A R
 H P E R I O D I C T A B L E M U I H T I L E V P
 C A L K A L I N E E A R T H M E T A L S P R U K

Alkaline earth metals

alkali metals

Reactivity

Strontium

Potassium

Group 3A

Francium

Calcium

Indium

Cesium

Boron

Valence electrons

Atomic number

Properties

Magnesium

compounds

Group 1A

Rubidium

Lithium

Radium

Sodium

group

periodic table

Atomic mass

Aluminium

Beryllium

Thallium

Group 2A

Gallium

element

Barium

symbol

name