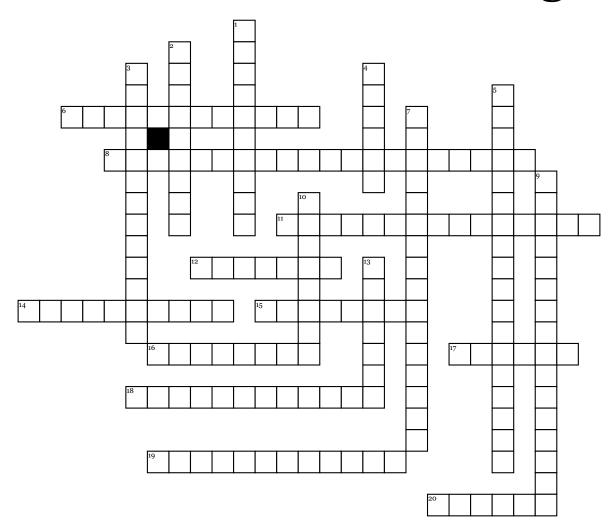
Name:	Date:	Period:

Periodic Table Challenge



Across

- **6.** Who arranged the elements according to increasing in atomic numbers?
- **8.** Have some of the characteristics of the transition metals.
- 11. Elements found in Group
- **12.** Atom with no charge. (neutral)
- 14. Have properties of both metals and non-metals.
- **15.** Atom that is negatively charged.

- **16.** Other than period and groups, the table is divided into? electricity or heat very well.
- 17. A one or two letter abbreviation.
- 18. Name for elements in Group 1.
- **19.** Equal to the number of protons in the nucleus.
- **20.** Each of the columns are called?

Down

1. Colorless gases that are extremely unreactive.

- 2. Not able to conduct
- 3. Organizational system for elements.
- **4.** Atom is positively charged.
- 5. Elements not found free in nature.
- 7. Who created the periodic
- **9.** Elements in the center of the periodic table.
- 10. Elements found in Group
- **13.** Each of the rows are called?