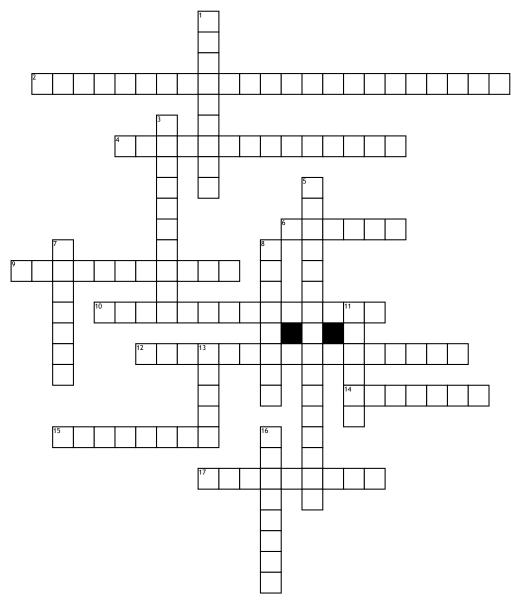
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Periodic Table and Properties



Across

- **2.** Matter is not created or destroyed
- **4.** Formation of a new substance with different properties; cannot be undone or changed back
- **6.** Shiny, Malleable, ductile, and good conductors
- **9.** Number placed in front of a chemical formula in a chemical equation; represents the number of molecules of that substance
- **10.** Change from ice to water, or water to steam

- **12.** Property of matter that can be observed without changing the composition or identity of the matter.
- **14.** Substances resulting from a chemical reaction
- **15.** How many groups are in the Periodic Table
- **17.** Dull, Brittle, and poor conductors

<u>Down</u>

- 1. Number written to the right and slightly below an element in a chemical formula, representing the number of atoms for that element.
- **3.** Properties of both metals and non-metals

- **5.** Characteristic of a substance that describes how it combines with other substances to form new ones.
- **7.** horizontal rows on the periodic table
- **8.** Substances that are combined and changed during a chemical reaction
- **11.** Vertical columns on the periodic table
- 13. How many periods are in the periodic table
- **16.** A substance made of two or more elements