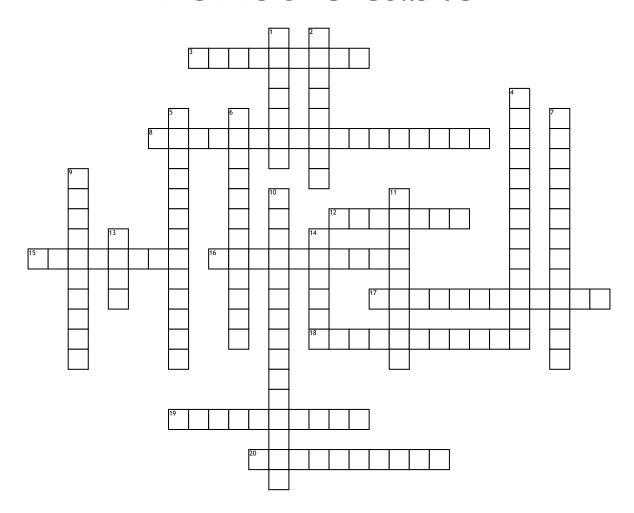
Name:	Date:	

Periodic table



Across

- **3.** not able to conduct heat or electricity
- **8.** this property describes an atoms ability to attract an electron
- **12.** arranged the elements according to increasing atomic numbers
- **15.** a negative charged particle
- **16.** discovered the existence of an atomic nucleus
- 17. group 1, most reactive to all metals, 1 valence electron, shiny'

- **18.** happy/inert elements(full outer shells colorless gasses that are extremely unreactive
- 19. total number of protons and neutrons in the nucleus
- **20.** have properties of both metals and non metals

Down

- 1. particle with no charge
- **2.** group 17, means SALT FORMER
- **4.** center area of atom composed of protons and neutrons
- **5.** describes where electrons go

- **6.** number of protons in a atom
- **7.** an organizational systems for elements
- **9.** the scientist that discovered the nucleus
- **10.** a tube that has two pieces of metals called electrodes
- **11.** proposed arranging elements by weights and properties
- 13. the basic unit of a chemical element
- 14. positive charged particle