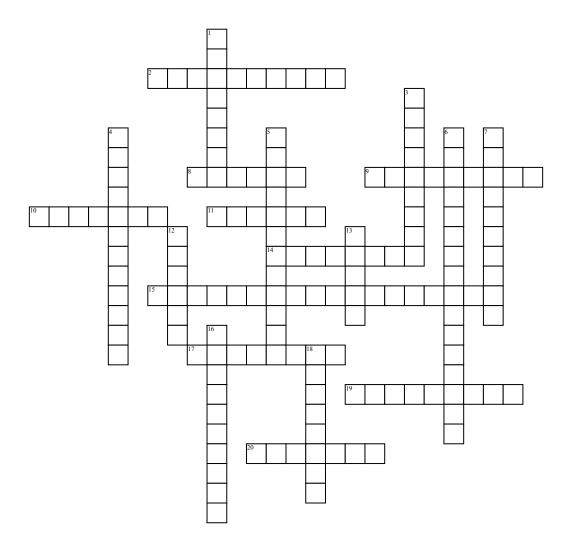
Name:	Date:	Period:

## Periodic table



## **Across**

- **2.** any of the series of fifteen metallic elements from lanthanum to lutetium
- **8.** Tells you how many energy levels there are
- **9.** Found in between metals and nonmetals
- **10.** fluorine, chlorine, bromine, iodine, and astatine. Group 17
- **11.** Alkali Metals, Halogens, Noble Gases, etc.

- **14.** A gaseous element that is not reactive
- 15. protons, neutrons, and electrons
- 17. Protons+?=Atomic Mass
- **19.** Used to model how many protons, neutrons, and electrons there are
- **20.** Atomic number tells you how many there are

## **Down**

- 1. Actinum through lawrencium
- **3.** Have a negative charge

- **4.** Determined using the periods
- **5.** Tells you how many protons there are
- **6.** Determined using group number
- 7. Protons + neutrons
- **12.** Columns. Also tell you number of valence electrons
- **13.** Found on left side of the periodic table
- **16.** How reactive a substance is
- **18.** Found on the right side of the periodic table

## **Word Bank**

Electrons	Halogen	atomic number	Actinide	Groups
Nonmetal	valence electrons	Atomic mass	Period	Noble Gas
Reactivity	Protons	Bohr Model	Subatomic particles	Neutrons
lanthanide	Metalloid	Metal	Energy levels	Family