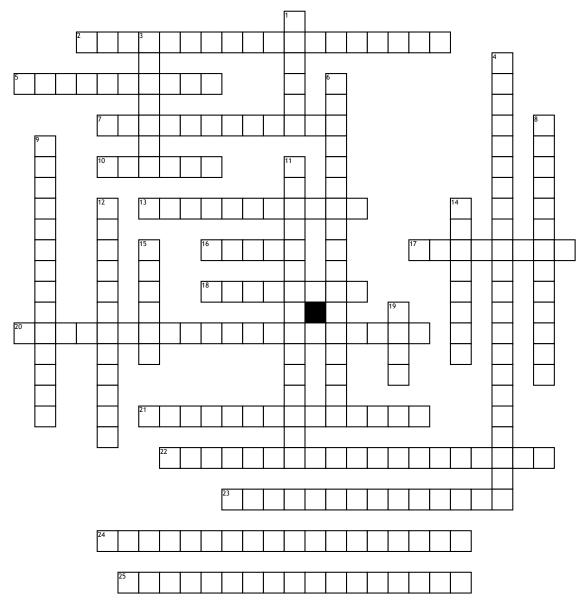
Name:	Date:
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Properties of Water



Across

- 2. Any property that is measurable
- 5. amount of substance that dissolves in a given quantity
- 7. amount of heat needed to increase the temperature of 1g
- **10.** The volume of a cube measuring 10 centimeters on each edge
- **13.** Compound that conducts an electric current when it is in an aqueous solution
- **16.** A condensable gas
- **17.** Concentration of solute in a solution expressed as the number of moles of solute
- **18.** Process of preparing a less concentrated solution from a more concentrated solution

- **20.** Solution containing a large amount of solute
- **21.** an inward force that tends to minimize the surface area of a liquid
- **22.** a solution that contains less solute than a saturated solution
- **23.** Solution that contains a small amount of solute
- **24.** The amount of energy needed to convert s liquid to a gas
- **25.** Covalent bonds between atoms in which the electrons are shared unequally

<u>Down</u>

- 1. Measure of the space occupied by a sample of matter
- 3. The dissolving medium in a solution

- **4.** A solution that contains more solute than it can theoretically hold at a given temperature
- Solution containing the maximum amount of solute for a given amount of solvent
- **8.** Measurement of the amount of solute the is dissolved in a given solvent
- **9.** Compound that does not conduct an electric current in molten state
- 11. 2 hydrogen atoms for single polar bonds with 1 oxygen atom
- **12.** Attraction between negative regions to positive regions
- **14.** Homogenous mixture; consists of solutes dissolved in a solvent
- **15.** Dissolved particles in a solution
- **19.** Amount of a substance that contains 6.02x10 to the 23rd power