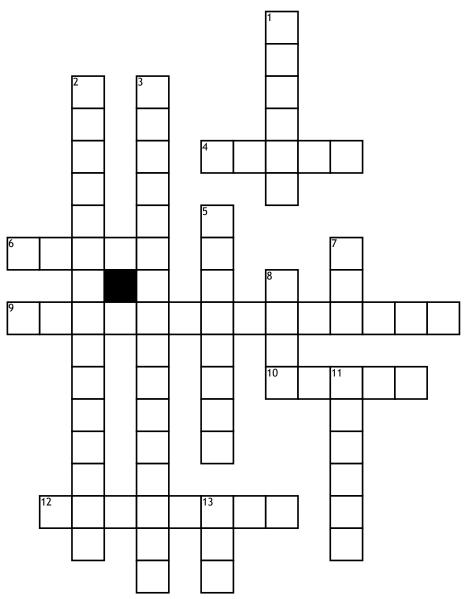
Properties of Water - Unit 2



Across

- 4. A substance on Earth that is found in all three phases/states (solid, liquid, gas)
 6. The positive and negative ends of a water molecule make it a molecule
- **9.** Because of strong hydrogen bonds and cohesion the outside of liquid water acts like a shield or skin allowing things to float
- 10. This state of water has its molecules form special patterns due to bonds made when molecules stop moving (empty space, expands)
- **12.** When water molecules are attracted to other water molecules

<u>Down</u>

- 1. A substance that is to be dissolved in a solution by a solvent
- 2. When water travels up/through a substance (against the force of gravity

- **3.** Water dissolves more substances than any other liquid (due to its polarity)
- **5.** When water molecules are attracted to other substances
- **7.** State of water where molecules are far apart and moving fast
- 8. Ice (solid water) is _____ dense than liquid water

11. State of water where molecules are densely packed but move about freely

13. Solid water is known as