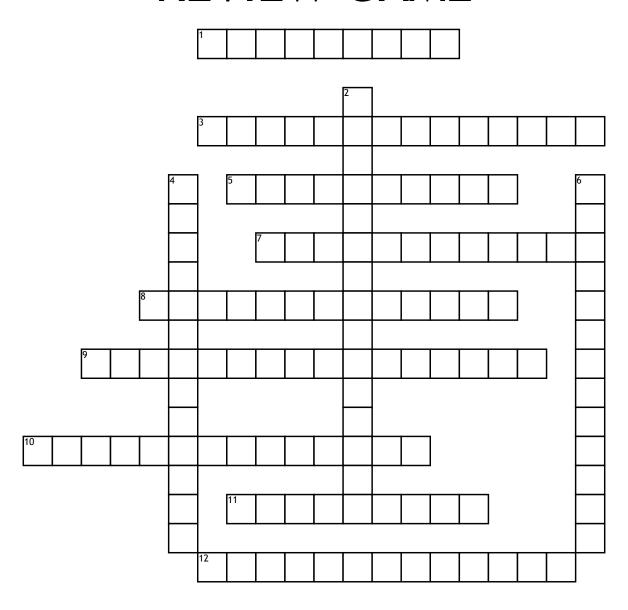
REVIEW GAME



Across

- 1. any substance that can donate a pair of electrons to form a covalent bond
- **3.** any acid that contains three ionizable protons (hydrogen ions)
- **5.** a substance that can act as both an acid and a base
- 7. the positive ion formed when a water molecule gains a hydrogen ion
- **8.** the particle that remains when an acid has donated a hydrogen ion

- **9.** the point in a titration where the number of moles of hydrogen ions equals the number of moles of hydroxide ions
- **10.** any solution in which the hydrogen-ion concentration is greater than the hydroxide-ion concentration
- **11.** any substance that can accept a pair of electrons to form a covalent bond
- **12.** any solution in which the hydroxide-ion concentration is greater than the hydrogen-ion concentration

Down

- **2.** any acid that contains one ionizable proton (hydrogen ion)
- **4.** any acid that contains two ionizable protons (hydrogen ions)
- **6.** the particle formed when a base gains a hydrogen ion