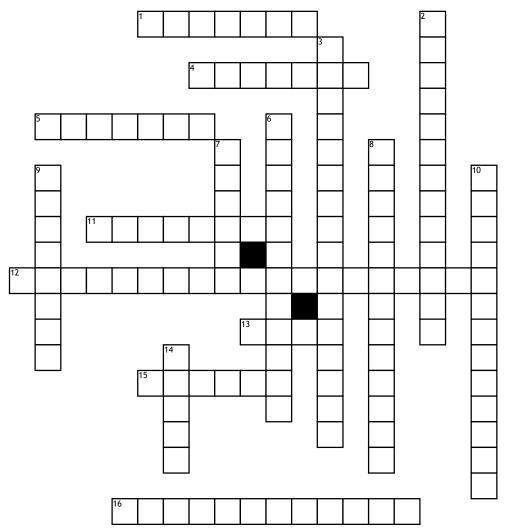
Name: _____ Date: _____

Radioactivity



Across

- **1.** The splitting of an atomic nucleus into two smaller parts
- **4.** Contains protons and neutrons in an atom
- **5.** A variation of an element with the same number of protons but a different number of neutrons
- **11.** The time required for one half of a sample of radioactive sample of a radioisotope to decay
- **12.** Nuclear radiation that occurs naturally in the environment

- **13.** An electron emitted by an unstable nucleus
- 15. The fourth state of matter
- **16.** Any atom containing an unstable nucleus

Down

- **2.** The conversion of atoms of one element to atoms of another
- **3.** The charged particles and energy that are emitted (given off) from the nuclei of radioisotopes
- **6.** Atoms of one element can change into atoms of a different element

- 7. A process in which the nuclei of two atoms combine to form a larger nucleus
- 8. The process in which an unstable atomic nucleus emits charged particles & energy
- **9.** A penetrating ray of energy emitted by an unstable nucleus
- **10.** A chain of fission reactions triggered by neutrons released during the initial fission of a nucleus
- **14.** A positively charged particle made up of two protons & two neutrons

Word Bank

Fusion Alpha Half life Isotope
Radioactivity Beta Background radiation Gamma ray
Fission Nucleus Nuclear radiation Plasma

Chain reaction Radioisotope Nuclear Decay Transmutation