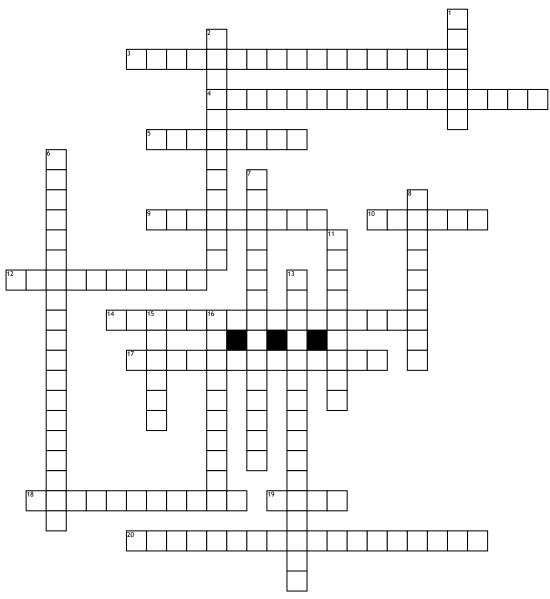
Name:	Date:
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Science 7-2



Across

- **3.** when one compound combines with another
- **4.** when a more active element takes the place of another element and sets the less active one free
- **5.** substances you end with
- 9. when 2 or more elements or compounds combine to make a complex compound
- **10.** in a chemical equation, the arrow means
- **12.** matter can enter from or escape to the surroundings

- 14. evidence for this reaction
- **17.** When a compound breaks into its component parts
- **18.** Indicates how many molecules or atoms of each reactant and product are involved in a reaction
- **19.** conservation of mass
- **20.** when energy is released **Down**
- 1. Cannot be created or destroyed
- **2.** matter is not allowed to enter or leave
- 6. energy is absorbed

- **7.** A combination of symbols that represents the elements present in the formula
- 8. substances you start with
- **11.** Indicates the ratio of elements in a compound
- **13.** uses formulas and symbols
- 15. reactions require
- **16.** occurs when a substance reacts with oxygen to produce energy