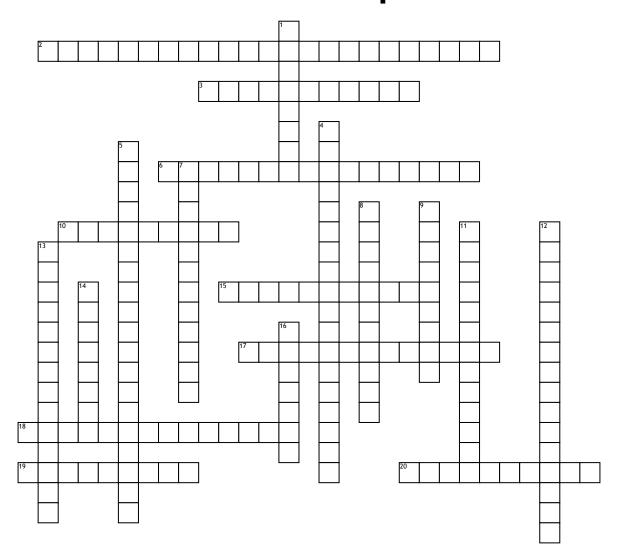
Name:	Date:
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Science Chapter 4



Across

- **2.** Law that states during a chemical reaction, matter is not created or destroyed
- **3.** When one element replaces another in a compound
- **6.** Minimum amount of energy needed to start a chemical reaction
- **10.** When two or more elements combine to make more complex substances
- **15.** Number placed in front of a chemical formula
- **17.** Amount of substance in a given volume

- **18.** A change that alters the form or appearance of a substance but does not change it to any other substance
- **19.** Material used to decrease rate of a chemical reaction
- **20.** Matter can enter or escape surroundings

Down

- 1. Increases reaction rate by lowering the activation energy needed
- **4.** The change that releases energy
- **5.** A change that absorbs energy
- 7. Matter can not enter or leave

- **8.** A solid that forms from liquids during a chemical reaction
- **9.** Substance that undergo the chemical changes
- 11. Occurs when compounds break down into simpler products
- **12.** A way to show chemical reactions using symbols
- **13.** A change in matter that produces one or more new substances
- 14. The new substances that form
- **16.** Cells in your body contain thousands of biological catalysts called