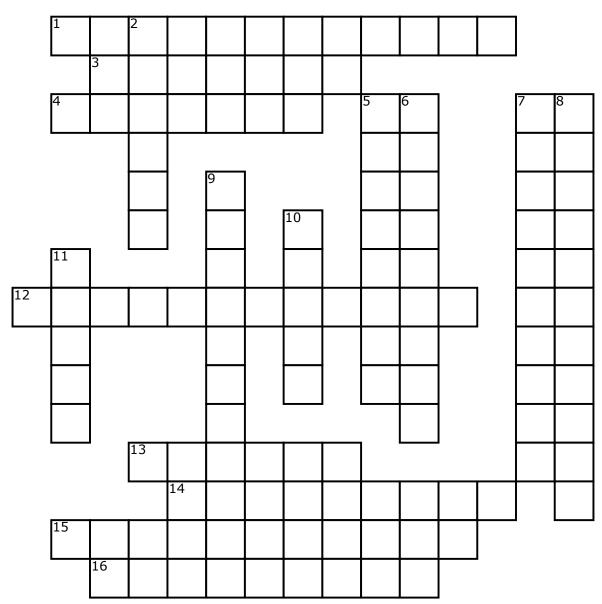
Name:	Date:	Period:	

Solubility Crossword



Across

- **1.** Breaking up an compund into its ions Ex. HCl -> H + Cl
- 3. Capable of being disolved
- **4.** Part of a solution that is present in the largest amount
- **12.** How materials conduct electricity
- **13.** An ion with a positive charge
- **14.** When no more of s subtance can dissolve or mix into the solution
- **15.** When more of a substance can dissolve in a solution

16. Compoud comsiting or 2 non-metals. Ex. HCl (_____ compoud)

Down

- **2.** Substance that is dissolved in another.
- Moles/Litre
- **6.** Process that results in a negative or positively charged atom or molecule (this is due to losing or gaining an electron)
- **7.** How much of a substance can dissolve at specific conditions.

- **8.** A solution that conducts electricity
- **9.** The formation of a solid in a solution
- **10.** A group of atoms with a negative charge
- **11.** A compund made up of a metal and a non-metal. Ex. NaCl (_____ compound)