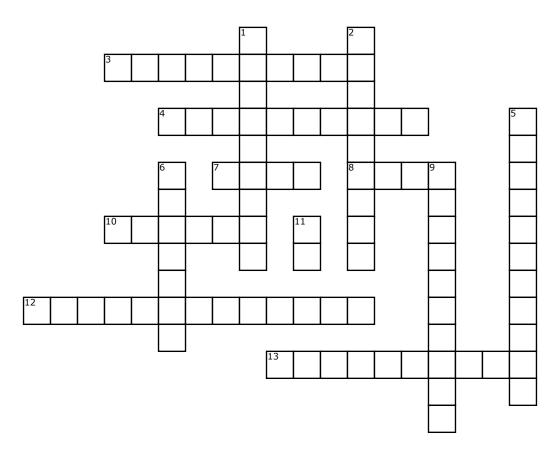
Solutions Acids and Bases



Across

- **3.** some molecule compounds
- **4.** amount of solute that dissolves in given amount of solvent of particular temp.
- **7.** ionic compound that can form by neutralization reaction
- **8.** compound that produces hydronium ion in solution

- **10.** substance that gets dissolved
- **12.** two or more component (solvent and H20)
- **13.** molecule to split

Down

- **1.** solution that contains as much solute as solvent can hold a given temp.
- **2.** Visible sign by color change

- **5.** solution that contains more solute then it canthan it can normally at a given temp.
- **6.** Substance that does the dissolving
- **9.** molecule compounds
- **11.** scale of acidity starting from ZERO to FOURTEEN

Word Bank

pH Ionization Salt Dissociate Solubility
Dispersion Indicator Solute Solvent saturated

Acid Concentration Unsaturated