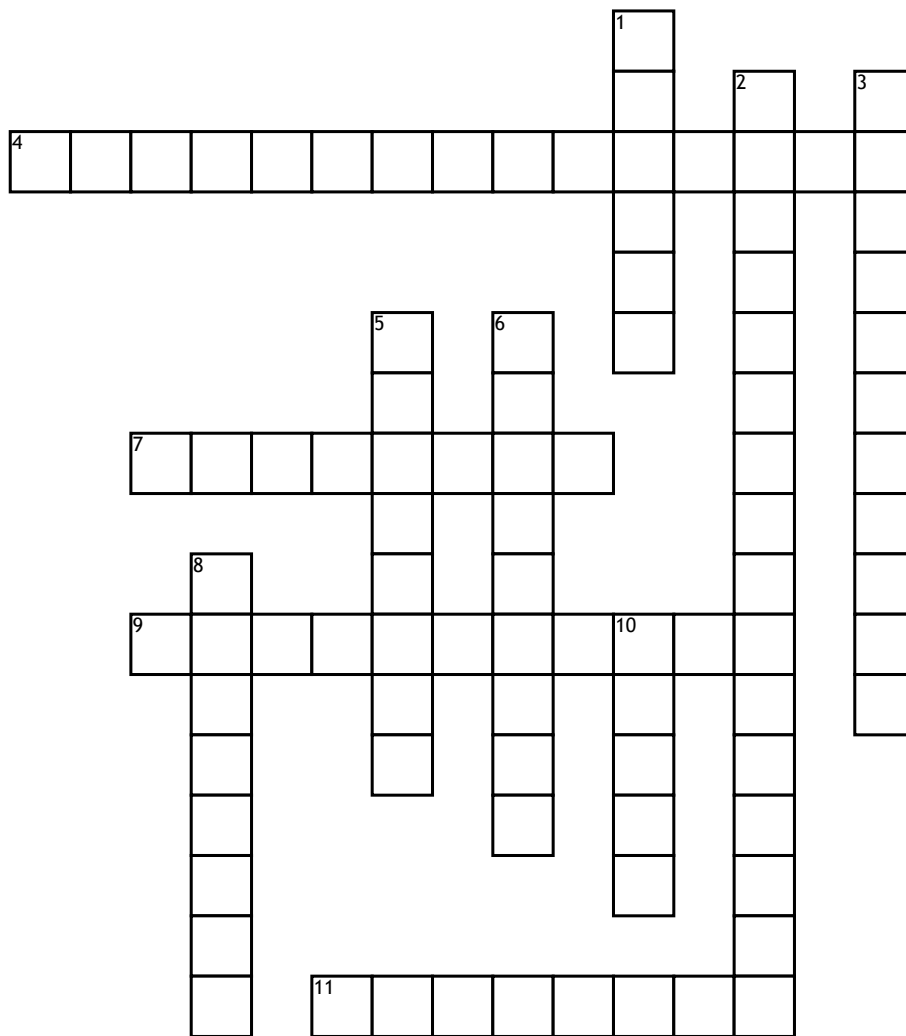


Name: _____ Date: _____ Period: _____

Solutions Review Game



Across

4. What value can you use when calculating concentration?
 7. The number of moles of solute per kilogram of solvent is called what?
 9. Properties that depend upon the concentration of solute molecules or ions
 11. What does a solute dissolving in a solvent create?

Down

1. A substance that is dissolved in a solvent is called a _____

2. What things you can do to make a solute dissolve faster?
 3. Fewer solutes than solvent in the solution is called _____
 5. Moles of solute over liters of solution is how you calculate what?
 6. A solution in which no more solute can be dissolved in the solvent is called?
 8. What type of compound doesn't dissolve in water?
 10. A compound dissolved in water that make an electrically conductive solution is called what?

Word Bank

Colligative

Molality

Molarity

Solute

PartsPerMillion

MixHeatBreakdown

Covalent

Solution

Ionic

Saturated

unsaturated