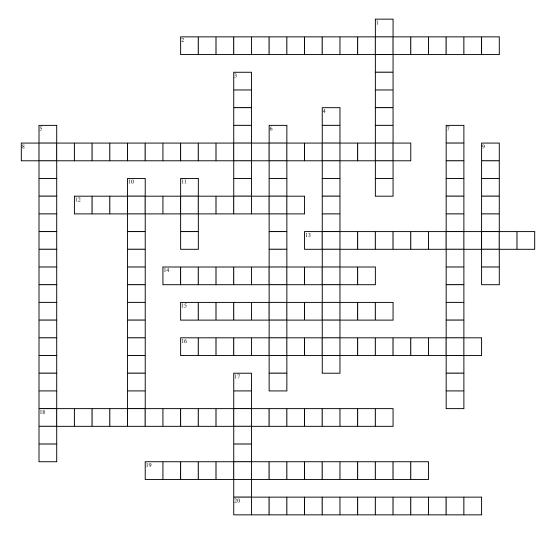
States Of Matter



Across

- 2. acts on subatomic particles that are externely close together and overcomes the electrostatic repulsion among protons
- **8.** a series of nuclear reactions that begins with an unstable nucleus and results in the formation of a stable nucleus
- 12. the combining of atomic nuclei
- **13.** a reaction in which an atoms atomic number is altered
- **14.** a sample that is massive enough to sustain a chain reaction
- sustain a chain reaction

 15. isotope of atoms with unstable nuclei
- **16.** radiation energetic enough to ionize matter with which it collides

- **18.** the process which involves striking nuclei with high velocity particles
- **19.** radioactive decay process that involves the emission of a positron from a nucleus
- **20.** the splitting of a nucleus into fragments **Down**
- 1. the difference in mass between a nucleus and its component nucleons
- 3. protons and neutrons
- **4.** reactors able to produce more fuel than they use
- **5.** process of determining the age of an object by measuring the amount of a certain radioisotope remaining in that object

- **6.** the area on the graph with in all stable nuclei
- 7. the ability of radiation to pass through matter
- **9.** the time required for one half of a radioisotope to decay into its products
- **10.** occurs when the nucleus of an atom draws in a surrounding electrons and combines with a proton to form a neutron
- 11. a form of high energy electromagnetic radiation
- **17.** a particle with the same mass as an electron but positive charge

Word Bank

nuclear fusion band of stability mass defect half-life critical mass Positron emission breeder reactors x-ray nucleons eletron capture strong nuclear force penetrating power nuclear fission radioactive decay series transmutation induced transmutation Radioisotope positron radiochemical dating ionizing radiation