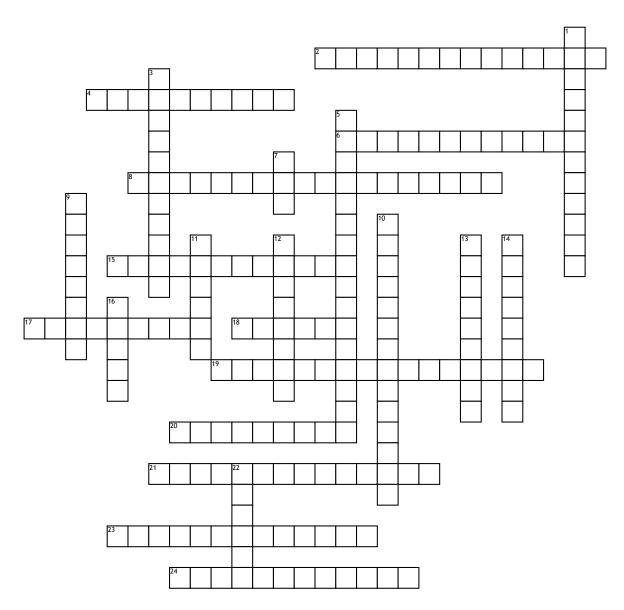
Name:	Date:
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States of Matter



Across

- 2.
- **4.** Discovered about buoyancy
- **6.** Solid phase turns to a liquid phase
- **8.** Amount of energy from liquid to gas
- **15.** Shows the temp change of water as heat is added
- 17.
- **18.** Matter of positive and negative charged particles **19.**

- **20.** Spreading of particles in a given volume
- **Ž1.**
- **23.** explanation of how particles in matter behave
- **24.** when pressure of the vapor is equal to the external pressure

Down

- 1. point when a solid begins to liquefy
- 3.
- **5.** Increase in size of substance when temp increases

- **7.** The particles are far apart and attract
- **9.** Force exerted per unit area
- 10.
- **11.** The Particles slide past each other
- **12.** Ability of a fluid to exert an upward force
- **13.** Resistance to flow by a fluid
- 14.
- **16.** The particles are closely packed together
- 22. The SI unit of pressure