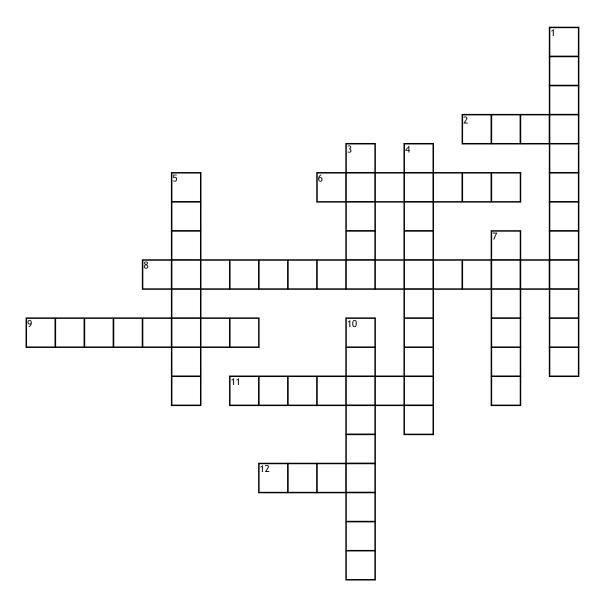
Name:	Date:	Period:

Structure of an ATOM



<u>Across</u>

- 2. The smallest unit of an element that maintains its properties.
- **6.** Two or more elements held together by physical forces
- **8.** What diagram shows how many valence electrons an element has?
- **9.** Elements combined chemically in a fixed ratio

- 11. The core of an atom, consisting of the protons and neutrons.
- **12.** Last name of the scientist who discovered structure of an atom.

<u>Down</u>

- **1.** Number of protons/ Number of electrons
- **3.** The maximum number of electrons in the second and third shell

- **4.** Number of protons + Number of neutrons =
- **5.** Two or more atoms held together by chemical bonds and act as a unit
- **7.** What in the nucleus carries a positive charge?
- **10.** Negatively charged particles that revolve around the nucleus.