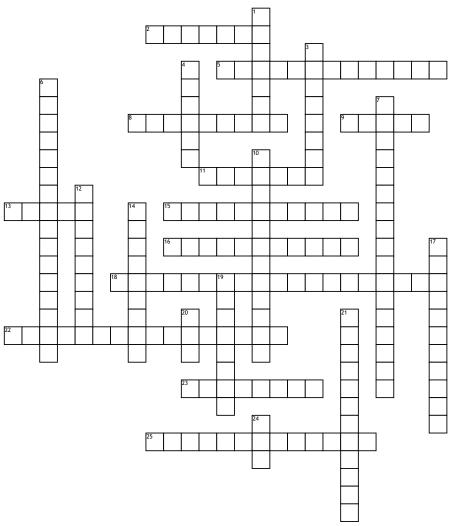
## The Table and Atoms



## Across

- 2. positivly charged
- **5.** wene a atom decays it emits
- 8. all radioactive
- **9.** evry verdical column in the Periodic Table of Elements
- **11.** made of protons and nutrons
- **13.** a electricity conducting group of elaments
- **15.** rare earth metals
- **16.** atom with low amounts of electrons
- **18.** reactive to group 17
- **22.** good cunducters
- 23. denoted by color

**25.** shows the probability raight for the placing of a sertan partical

## Down

- **1.** a atom with alot of weight in neutrons
- 3. very reactive with groups 1&2
- **4.** a horisontal line of elaments in the periodic table of elaments
- **6.** wene a atom has to meny protons it starts to
- 7. very stable, color less and odorless
- 10. very reactive with group 17
- **12.** negitively charged, cercals the nucleus

- **14.** has proprtes of metal, and nonmetals
- 17. too meny elecrons inan atom makes it
- 19. no charge
- **20.** \_\_\_\_\_ mass unit
- 21. the amount of protons an atom has
- **24.** atom with more or less then normal amonts of electrons

## **Word Bank**

TYOTA BAILIT				
transition metals	metalloid	radioactive decay	electron cloud	halogens
period	positive ion	atomic number	actinides	lanthanides
ion	ratioactivity	nucleus	alkali metals	protons
noble (inert) gases	group	AMU	nonmetal	neutrons
metal	alcaline earth metals	electrons	negitive ion	isotope