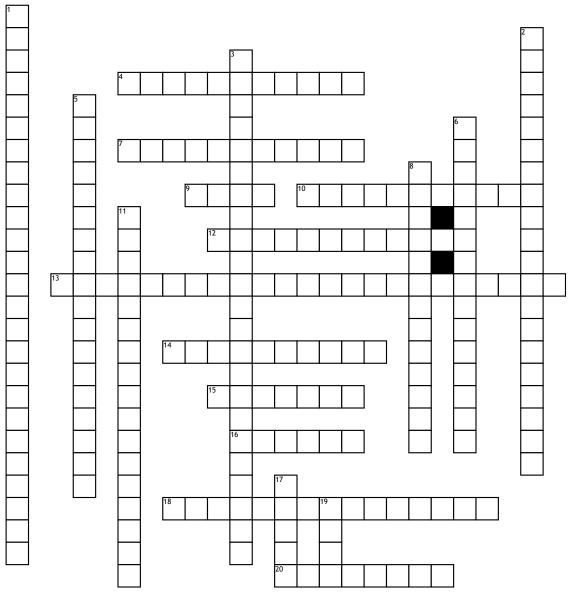
Thermochem Vocab



Across

- **4.** A reaction that sucks energy in from around it
- **7.** Device for measuring the heat involved in a chemical reaction
- **9.** Indicates the direction of energy flowing
- **10.** Measuring the amount of heat transfered
- **12.** The heat required to raise 1 gram of a given substance by 1 degree (different for different substances)
- **13.** Heat necessary to vaporise one mole of a substance
- 14. A reaction that releases energy

- **15.** The energy needed to raise 1 gram of water by 1 degree
- **16.** What force is needed to do work?
- **18.** What is the branch of chemistry that deals with how heat is absorbed/emitted during reactions?
- **20.** Total heat content of a system **Down**
- 1. Energy can be neither destroyed nor created
- 2. The Enthalpy change when 1 mole of a substance is completely burned in air at 298K and 1 bar of pressure.

- **3.** Form of potential energy from the chemical bonds of a molecule
- **5.** The Enthalpy change that occurs when one mole of a substance undergoes a chemical reaction
- **6.** Energy of Position
- 8. Energy of Motion
- **11.** How much energy is needed to melt 1 mole of a substance
- 17. The energy needed to raise 1 gram of a substance by 1 degree (equivalent to 1000 calories)
- **19.** The total energy from the vibration of all the molecules in a substance