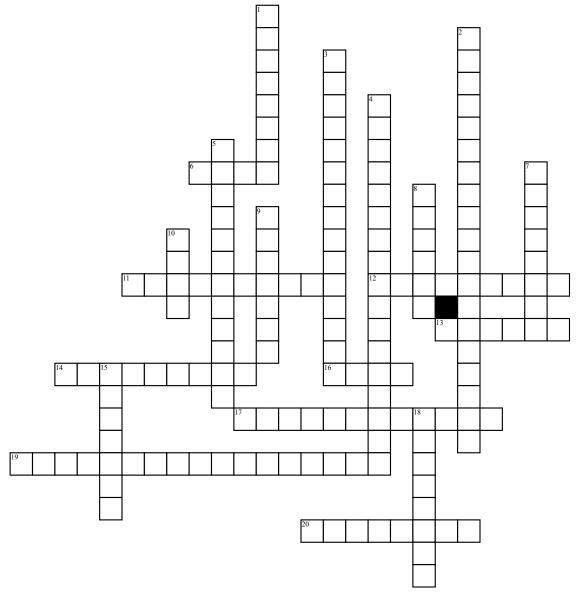
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Unit 1- Chemistry of Life



Across

- **6.** Positively and negatively charged atoms
- **11.** Total number of protons and neutrons in an atom
- **12.** Study of the composition of matter and the changes
- **13.** Amount of space an object occupies
- **14.** Formed when one or more electrons are transferred from one atom to another
- 16. Basic unit of matter
- **17.** Number of protons in the nucleus of an atom

- **19.** Inter molecular forces of attraction
- **20.** A substance formed by the chemical combination of two or more elements in definite proportions

Down

- **1.** Atoms of the same element that differ in the number of neutrons
- **2.** Have unstable nuclei that break down overtime
- **3.** Shows the kinds and number of each element in a compound

- **4.** Formed by chemical composition of two or more elements in definite proportions
- **5.** Formed when electrons are shared between atoms
- 7. A negatively charged particle (-)
- **8.** Anything that has mass and takes up space
- **9.** A pure substance that consists entirely of one type of atom
- **10.** Quantity of matter in an object
- **15.** Made up of protons and neutrons
- **18.** The smallest unit of most compounds