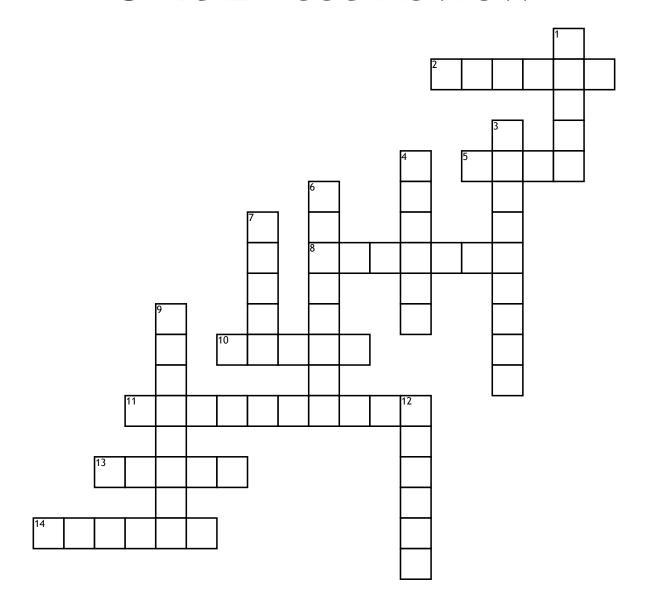
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Unit 2 Test Review



Across

- **2.** atom with positive charge
- **5.** number of electrons shared in a double bond
- **8.** electrons in highest energy level
- **10.** rule that atoms want 8 electrons in outer shell

- **11.** free ions dissociate and can carry electricity
- **13.** atom with negative charge
- **14.** melting point of ionic compounds

<u>Down</u>

- 1. melting point of covalent compounds
- **3.** combine to make covalent bonds

- **4.** oxidation number of Phosphorous
- **6.** bond where charges are shared
- **7.** bond where electrons are transferred
- 9. charges that attract
- **12.** sodium's atomic number