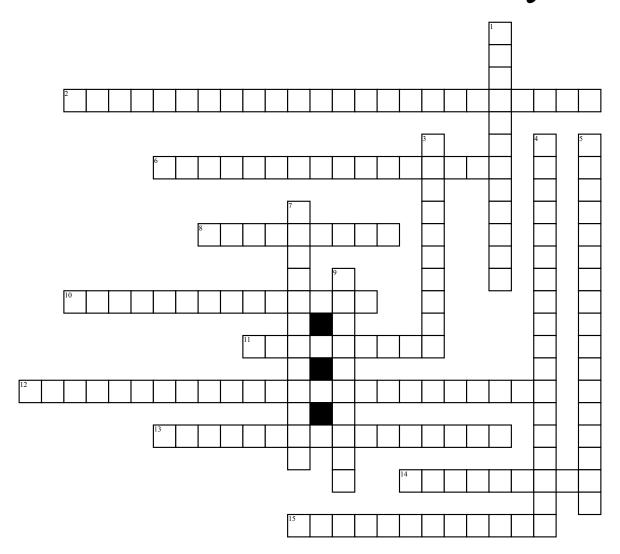
Name:	Date:
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Xander's Atom History



Across

- 2. states that a given chemical compound always contains its component elements in fixed ratio
- **6.** Electrons were in a space and surrounding a center mass
- **8.** Believed that there were only four elements
- **10.** Model of atom that concludes that electrons move in a directed way outside of the nucleus
- **11.** Discovered a negative particle called the electron

- 12. statement that when two elements combine with each other to form more than one compound, the weights of one element that combine with a fixed weight of the other are in a ratio of small whole numbers.
- **13.** Created the gold foil experiment
- **14.** Created the planetary model
- **15.** Theory that says matter is made of tiny particles that can't be subdivided

Down

- 1. describes the wave-like behavior of either one electron or a pair of electrons in an atom.
- **3.** Developed the Atomic Theory
- **4.** Matter Consist of fire, wind, earth, and water
- **5.** Explains that electrons move in a space not a path
- 7. The idea that matter can be divisible until a certain, specific point
- **9.** Created the Atomos Theory