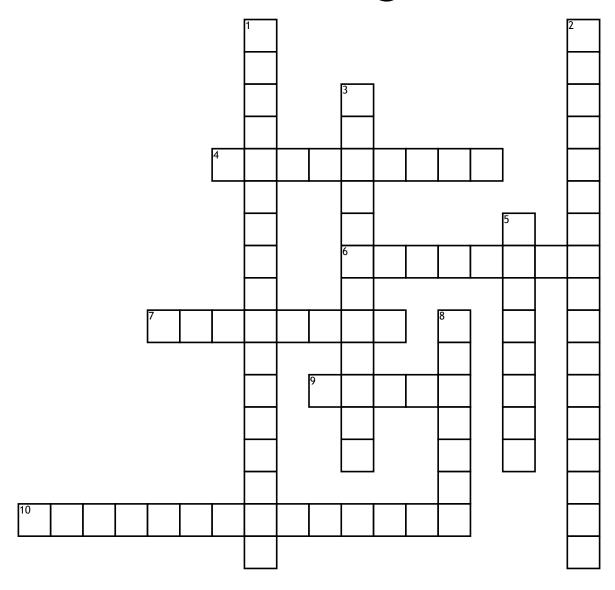
Name:	Date:
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## bonding



## **Across**

- **4.** Can be bent or hammered into thin sheets
- **6.** Between 2 nonmetals, electrons are shared
- **7.** Bonds allow atoms to obtain 8 valence electrons by losing, gaining, or sharing electrons
- **9.** Between a cation and an anion electrons are transferred from a metal to a non metal
- **10.** Combine to form a crystal lattice structure

## <u>Down</u>

- 1. losing valence electrons causes the metal atoms to become positive cations
- **2.** Atoms bond to form independent neutral particles called molecules, they can be small or large
- **3.** An electrostatic attraction between atoms or ions that holds compounds together
- **5.** Among atoms of a metal valence electrons form a se around positive metal cations
- **8.** Can be bent into a wire without breaking