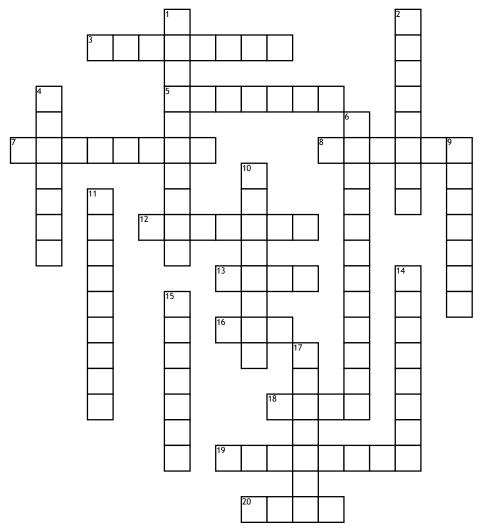
chapter 2- The chemistry of life



Across

- **3.** chemical combination of 2 or more substances
- **5.** measurement used to indicate the concentration of hydrogen ions (H+) in a solution
- **7.** a substance made by 2 or more elements
- 8. substance that is being dissolved
- **12.** substance in which a solute is dissolved
- 13. basic unit of matter
- 16. a charged atom

- 18. solutions with H+ hydrogen ions
- **19.** attraction between molecules of different substances
- **20.** solutions with OH+ hydrogen ions

Down

- 1. solution in which the material does dissolve but breaks into tiny particles that float whithin
- **2.** a compound formed by a covalent bond
- **4.** an atom that has more neutrons than it should

- **6.** a bond formed by the sharing of electrons
- **9.** a pure substnace consisting of one type of atom
- 10. negatively charged particle
- **11.** a bond formed by the transfering of electrons
- **14.** attraction between molecutes of the same substance
- **15.** physical combination of 2 or more substances
- **17.** the center of the atom; contains protons and neutrons.

Word Bank

solution	solute	molecule	mixture	suspension
acid	electron	covalent bond	compound	pH scale
ion	cohesion	adhesion	ionic bond	solvent
isotope	base	nucleus	element	atom