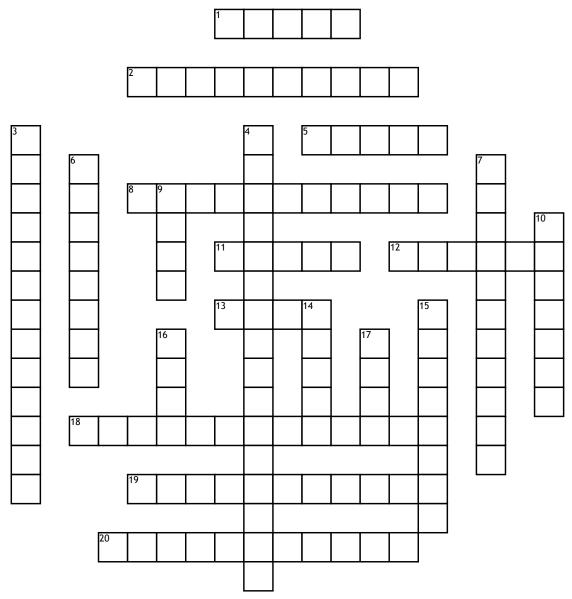
Name:	Date:	

## chem



## **Across**

- **1.** particle stopped by paper
- **2.** burning; H2O & CO2
- 5. metal-metal solution
- **8.** measure of the average kinetic energy
- **11.** vertical rows in the periodic table
- 12. space a gas occupies
- **13.** smallest unit of an element

- **18.** mass of a molecule held together of covalent bonds
- 19. shared pair of electrons
- **20.** mass of atoms held together by an ionic bond

## **Down**

- **3.** not used up in the reaction
- **4.** amount that could have been made
- **6.** unpaired share of electrons

- 7. amount actually made
- **9.** heat reaction in which energy is absorbed
- **10.** subatomic particle with charge of 0
- 14. basic unit of length
- **15.** amount of force applied
- **16.** counting unit for the amount of substance
- **17.** particle with charge of -1