

Name: _____ Date: _____

chemistry

S E S A G S T R U C T U R E S W R I P L F S L R
P Y I U Z N C N C F V B I W U N Q D A O R A R F
U P O S I T I V E M U E A D V T W X I R F A E Y
C C L Z T A Y D Y O K X U Q W G E O R O D A F P
I G S Q F Y W V C L L K U C J C X X S G R P L B
M L X U L R T I G E G E T E T O W J A G R P U N
O B X I X S Q C L C Y Y I P W V R E G S F K S P
T Z N N F E O F D U W J I P E A V J P T Q E O G
A I O K E S A C I L A I O N Y L T T N O F B J H
I Z L P W A K J U E K V R E U E Z T S D G B A F
D K T G V G Z L I O A F E G M N W T R O I G L N
G E N L K E F H J M D I T A G T Y T C W M D U U
T M E T A L L O I D S M A T P B V W T T O O M E
V N M E E B N U F J X T W I S O B L S H T C R T
E R E R L O E E K X Q U E V C N U D G H A T O R
O Y L Y G N P S L A T E M E M D F S R B I E F A
E L E C T R O N C O N F I G U R A T I O N T K L
T Z C X C J W A X J T W X A I Y K I Q P Z R G G
X G R H V C N T X L K U F P H L F X G J U U R Q
D I R H C N F R Z U C N G D N U O P M O C L N X
T Q B C S C I U Y M Q F U L N O R T E U N E K N
L I Q U I D S V W D Y L I M A F F S O L I D S L
U X K B J P R O T O N F U W I U A Z W H S Y T R
Q J L L J P E R I O D I C T A B L E Y U G N E E

electron configuration	periodic table	covalent bond	noble gases
metalloids	octet rule	structure	compound
diatomic	molecule	negative	positive
two dots	element	formula	liquids
nuetral	nuetron	family	metals
proton	solids	sulfer	gases
pairs	water	atom	ion