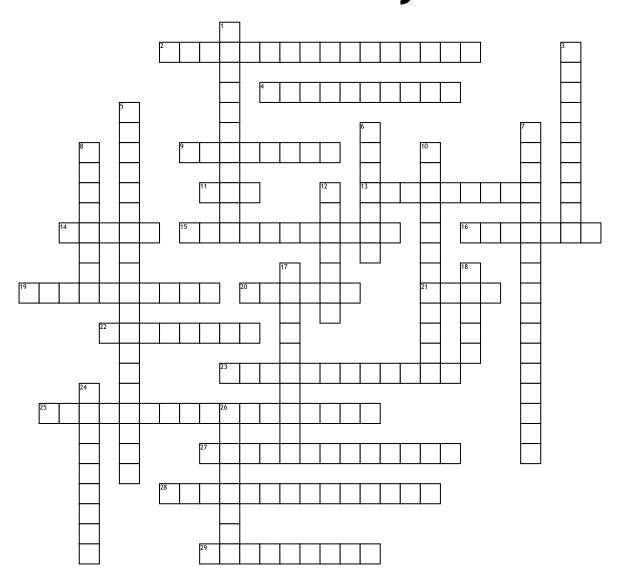
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## chemistry



## **Across**

- **2.** an electron the outer most energy shell of an atom
- **4.** a compound consisting of a nucleoside linked to a phosphate group. Nucleotides form the basic structural unit of nucleic acids such as DNA
- 9. negative charge
- **11.** the total number of electrons is not equal to the total number of protons
- **13.** monomer of protein and are joined together by condensation
- 14. relating to, or using ions
- 15. tending to repel or fail to mix with water
- **16.** atoms of the same element can have different number of neutrons different versions of each element
- **19.** the chemical breakdown of a compound due to reaction with water

- **20.** an odorless tasteless white substance occurring widely in plant tissue and obtained chiefly from cereals and potatoes
- 21. made up of subatomic particles
- 22. a colorless, odorless, highly flammable
- gas, the chemical element of atomic number 1
- 23. share electrons (strongest bond)
- **25.** a measure of the tendency of an atom to attract a bonding pair of electrons
- **27.** phospholipid, phosphatidylcholine, monoglycerides and diglycerides
- 28. monomer of a carbohydrate
- **29.** a single strand of nucleotides, and it occurs in a variety of lengths and shapes
- 1. a chemical bond formed between two molecules when the carboxyl group of one molecule reacts with the amino group of the other molecule, releasing a molecule of water
- **3.** consisting of a nitrogen atom attached by single bonds to hydrogen atoms

- 5. overall charge of the atom is zero
- 6. made up of mainly carbon
- **7.** a type of chemical reaction that involves a transfer of electrons between two species
- **8.** covalent bonds are a type of bond that occurs when two atoms share a pair of electrons with each other.
- **10.** water that collects as droplets on a cold surface when humid air is in contact with it
- **12.** number of protons and electrons in an element
- 17. major component of all lipids (a long hydrogen chain)
- **18.** covalent bond between two atoms where the electrons forming the bond are unequally distributed.
- **24.** an organic compound ad makes most of the plants cell wall
- **26.** a substance deposited in bodily tissues as a store of carbohydrates. It is a polysaccharide that forms glucose on hydrolysis