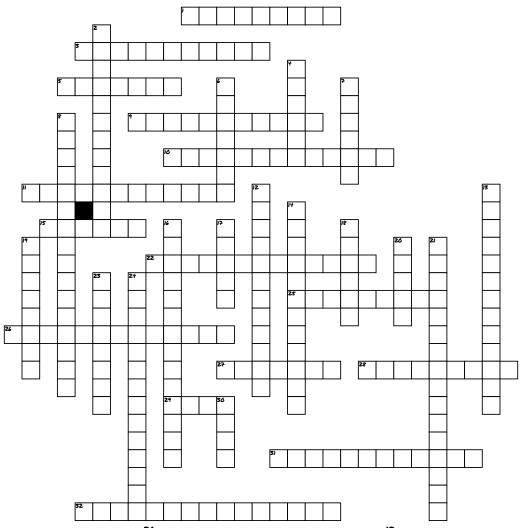
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## SCIENCE VOCAB



## ACROSS

- 1. TO BEND INTO A WIRE
- S. THE NATURE OF SOMETHING'S INGREDIENTS OR CONSTITUENTS; THE WAY IN WHICH A WHOLE OR MIXTURE IS MADE UP
- S. WHAT IS BEING DISSOLVED
- **9.** A MIXTURE WHERE THE COMPONENTS THAT MAKE UP THE MIXTURE ARE UNIFORMLY DISTRIBUTED THROUGHOUT THE MIXTURE.
- 16. MIXTURE HAVING A NON-UNIFORM COMPOSITION. THE COMPOSITION VARIES FROM ONE REGION TO ANOTHER
- 11. TURNS TO A GAS
- 15. ANYTHING THAT HAS VOLUME AND MASS
- 22. THE DATA STRUCTURE THAT HOLDS ELEMENTS
- **25.** A SECTION ON THE TABLE THAT MOSTLY IS MADE UP OF GASSES

- **26.** Anything composed of a single type of material.
- 27. THE DEGREE OF COMPACTNESS OF A SUBSTANCE
- 28. THE SECTION THAT IS SEMICONDUCTORS
- 29. THE BASIC UNIT OF A CHEMICAL ELEMENT.
- 3). THE HEAT REQUIRED TO RAISE THE TEMPERATURE OF THE UNIT MASS OF A GIVEN SUBSTANCE BY A GIVEN AMOUNT
- 32. Another name for the group number

## DOWN

- 2 THE POINT AT IT IS ANT THE POINT OF VAPOR
- 4. WHAT IS DISSOLVING
- 6. ARE DEFINED BY THE NUMBER OF PROTONS THEY POSSESS.
- 7. A HORIZONTAL ROW OF THE PERIODIC TABLE.
- 8. A PROPERTY OR CHARACTERISTIC OF A SUBSTANCE THAT IS OBSERVED DURING A REACTION

- 12 SOLID TO A LIQUID
- 13. COLD POINT
- 14. The point at which is changing a solid to a liquid
- 16. WHEN IT IS CHANGE INTO A NEW SUBSTANCE
- 17. THE VERTICAL LINE ON THE PERIODIC TABLE
- 18. THE AMOUNT OF SPACE BEING TAKEN UP
- 19. TWO OR MORE ELEMENTS CHEMICALLY COMBINED
- 20. A SECTION ON THE TABLE THAT IS SHINNY AND CONDUCTIVE
- **21.** A CHARACTERISTIC OF MATTER THAT CAN BE OBSERVED AND MEASURED WITHOUT CHANGING THE CHEMICAL IDENTITY OF THE SAMPLE.
- 23. A HETEROGENEOUS MIXTURE
- 24. WHEN THE APPEARANCE IS CHANGED
- 30. HOW MUCH SOMETHING WEIGHS

## WORD BANK

PHYSICAL CHANGE COMPOUND
DENSITY BOILING POINT
CHEMICAL PROPERTY MELTING POINT
GROUP SOLUTION
METALLOID SOLUTE
DUCTILITY COMPOSITION

ATOM METAL
HOMOGENEOUS SPECIFIC HEAT
MASS VOLUME
NONMETALS MELTING POINT
FREEZING POINT PHYSICAL PROPERTY

PERIODIC TABLE HETEROGENEOUS
ELEMENT PERIOD
PURE SUBSTANCE CHEMICAL CHANGE
VALENCE ELECTRON BOILING POINT
SOLVENT MATTER